

### Synthesis and investigation of plasma modified mesoporous silica nanostructured properties obtained from cereals husk and their application in drug delivery

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## Outline

- Introduction
- Literature Review
- Motivations
- ➢ Research Gap
- Thesis Objectives
- Research Methodology
- ➤ Time Table

Cancer is when abnormal cells divide in an uncontrolled way. Some cancers may eventually spread into other tissues



- Chemotherapy has side effects to body
- Chemotherapy drugs can destroyed by enzymes
- > Tis method is not private







#### > What's the solution?



Paul Ehrlich 1991

#### Magic bullet, or shot in the dark?









Mesoporous Silica Nanoparticle (MSN)





#### **Features**

- > Stable structure
- > Uniform particle size
- Large surface area
- ➢ High pore volume
- > Tunable pore diameters
- Covered by silanol groups
- MSN is accepted by the FDA
- ➢ Biocompatible

- Ability to encapsulate a variety of therapeutic agents such as drugs, fluorescent dyes & MRI active chelate
- High drug loading capacity
- surface modifications with functional groups
- Controlled release with stimuli-responsive systems
- Sustained release



Drug
 Fluorescent dye
 MRI active chelate
 Surface charge tuning







- Lignocellulose: 72-85 %
- ➢ Silica: 15-28 %

Silica Nanoparticles synthesis from cereal husks



 $\succ$  SiO2 + NaOH → Na2SiO3 + H2O

Mesoporous Silica Nanoparticles modification

→ Post synthesis grafting



Mesoporous Silica Nanoparticles modification

- Post synthesis grafting
- → Co-condensation



### Plasma

Plasma can be defined as a quasi-neutral gas of charged and neutral particles characterized by a collective behavior



### **Species in Plasma**

- Electrons
- Positive ions
- Negative ions
- Electronically excited atoms and molecules
- Vibrationally and Rotationally excited molecules
- Active Radicals
- > UV ultraviolet light photons
- Neutral species



Depending on their energy level, temperature and pressure, plasmas are usually classified as:

|             |   | 277N   |  |  |  |
|-------------|---|--|--|--|--|
|             | Thermal Plasma                                    | Cold Plasma                                    |  |  |  |
| Temperature | <ul> <li>High electron temperature</li> </ul>     | <ul> <li>High electron temperature</li> </ul>  |  |  |  |
|             | <ul> <li>High gas temperature</li> </ul>          | <ul> <li>Low gas temperature</li> </ul>        |  |  |  |
| Pressure    | Atmospheric or higher                             | Low ( < 0.1 bar) for most cases                |  |  |  |
| Appearance  | Filamentary inhomogeneous                         | Homogeneous                                    |  |  |  |
| Types       | <ul> <li>Plasma jet</li> </ul>                    | Glow   |  |  |  |
|             | <ul> <li>DC corona torch</li> </ul>               | <ul> <li>Radio Frequency</li> </ul>            |  |  |  |
|             | • arc   | <ul> <li>Microwave</li> </ul>                  |  |  |  |
| Uses        | <ul> <li>Ultra-fine particles spraying</li> </ul> | Modification or treatment of particles surface |  |  |  |
|             | <ul> <li>Sputtering</li> </ul>                    |  |  |  |  |

### **Glow discharge plasma**





### **Glow discharge plasma**

- Voltage: 1000 5000 V
- Current: 100 mA
- Bulk Temp: Room Temperature
- Electron Temp:10000 C
- ➢ Ion Temp:1000 C
- Pressure:0.1-10 torr
- Distance of Electrodes for Ar: 10 30 cm





### > Silica nanoparticles from rice husk

| In 1938 existence of silica in RH was Discovered      |  |  |  |  |
|---|--|--|--|--|
| -Synthesis amorphous                                  | Water pretreatment   |  |  |  |
| and crystalline silica from direct calcination of RH. | <ul> <li>-Water leaching of raw<br/>RH to remove adhering<br/>soil, dust, and some<br/>metal cations.</li> <li>-Water rinse can<br/>effectively remove most<br/>minerals except for K and</li> </ul> | Acid pretreatment  |  |  |
|   |  | -To completely remove<br>metal impurities and<br>carbonaceous residue.                     |  |  |
|   |  | - HCl<br>-H <sub>2</sub> SO <sub>4</sub>   |  |  |
|   | Ca in RH.  | -Organic acids, such as<br>acetic acid, oxalic acid,<br>citric acid and carboxylic<br>acid |  |  |

### Treatment and Modification by plasma



#### > Template removal

Min-Hao Yuan,2014, SBA-15 template removal by O2 glow discharge plasma to High Surface Area, High Silanol Density, and Enhanced CO2 Adsorption Capacity.

Yuan Liu,2009, MCM-41 template removal using dielectric-barrier discharge (DBD) plasma



### Plasma treatment effect on nanoparticles

Min-Hao Yuan,2014, SBA-15 treatment to improve chemical interaction between ATPES and silica

W.Yan,2012,SiO2 NPs treated by non-thermal plasma to disperse uniformly and form strong covalent bond with epoxy polymer matrix



### Direct surface modification

F.Taraballi,2012,Colagen amino and carboxyl functionalization by CO2 and N2/H2 plasma

P.Gandhiraman,2009,amine functionalization of cycloolefin by ATPES plasma



### Silica nanoparticles in drug delivery

- ✓ Physicochemical properties
- ✓ Simple surface modification

| Stimuli  | <b>Responsive Linker</b>                  | <b>Capping Agent</b>                         |  |  |
|--|---|--|--|--|
| Temperature  | Octadecyl (C <sub>18</sub> ) chains       | Paraffins                                    |  |  |
| Temperature  | PNIPAm                                    | PNIPAm                                       |  |  |
| Electric field   | 4(3-cyanophenyl)butylene dipolar molecule | -  |  |  |
| Magnetic field   | Hybridization of 2 ssDNA                  | $\gamma$ -Fe <sub>2</sub> O <sub>3</sub> NPs |  |  |
| Light  | Azobenzene derivatives                    | β-CDs  |  |  |
| pН   | Acetal linker                             | Au NPs                                       |  |  |
| pН   | Boronate ester                            | Fe <sub>3</sub> O <sub>4</sub> NPs           |  |  |
| pН   | Benzoic-imine bonds                       | Polypseudorotaxanes                          |  |  |
| Redox potential  | ox potential –S–S– se                     |  |  |  |
| Redox potential –S–S–  |   | PEG  |  |  |
| Enzymes  | Enzymes MMP-degradable gelatin Gelatin co |  |  |  |
| Enzymes β-galactosidase-cleavable oligosaccharide β-galacto-oligosac |   |  |  |  |

## Motivation

- Solution Solution Solution Structure and surface properties .
- Cereal husks are a excellent Biosources for MSNs synthesis and It is amazing to synthesis a valuable material from Bio waste.
- It is very interesting to use interdisciplinary science such as plasma to increase the efficiency of drug delivery systems.





Other Sites 46% Female incidence UK 2015 Lung 12%

Data source: GLOBOCAN 2018 Available at Global Cancer Observatory (http://gco.iarc.fr/) © International Agency for Research on Cancer 2018

**Motivation** 

### Research Gap

Mesoporous silica nanoparticles can green synthesis from cereal husks, while their properties are not investigate in drug delivery systems Effect of plasma on structure, physicochemical properties and modification of Mesoporous silica nanoparticles as nanocarrier are not investigate. In-Vitro and In-Vivo cytotoxicity of Mesoporous silica nanoparticles with cereal biosource and plasma modified are not investigate.

## **Thesis Objectives**

- > Bio-silica extraction from cereals husk such as rice husk and wheat husk.
- > Synthesis mesoporous silica nanoparticles from bio-silica precursor.
- Increase loading capacity to improve drug delivery ability.
- > Investigation on effect of plasma on structure and physicochemical properties of nanoparticles.
- > Surface modification of nanoparticles by glow discharge plasma.
- Investigation on drug delivery behavior of synthesized nanoparticles and evaluation drug loading and release profile.
- > Evaluation in-vitro and in-vivo cytotoxicity of synthesized nanoparticles

### Literature review

- Bio-silica from cereal husks
- Mesoporous silica nanoparticles
- Surface modification by plasma



### **Bio-silica extraction from cereal husks**



#### > Sol-Gel process to synthesis MSNs



Increase drug loading efficiency

Selective dissolution strategy

that dissolution of porous core starts from multiple nucleation sites, these small dissolved sites become larger overtime and interconnected with each other until complete core dissolution.



MSN modification by glow discharge plasma in drug delivery

- To compare physicochemical properties of NPs before and after plasma treatment
- To investigate the effect of plasma treatment on drug loading and release behavior
- To compare the effect of plasma on functionalization of MSNs between treated and untreated cases
- Direct surface modification by functional agents such as ATPES (the feed of reactor is functional agent)

### Characterizations

| No. | Analysis          |
|-----|-------------------|
| 1   | SEM, EDX          |
| 2   | FTIR Spectrometer |
| 3   | TGA               |
| 4   | BET               |
| 5   | XRD               |
| 6   | PORE VOLUME       |
| 7   | TEM               |

- > Drug delivery studies
  - Doxorubicin





### > Drug delivery studies

Doxorubicin loading and release analysis



### > Drug delivery studies

- > Doxorubicin release analysis
- ➢ Release profile detecting for 24 hours at pH 7.4, 6.5 and 5.5 and Temperature 37°C and 42°C



### > Cytotoxicity Analysis

- MTT assay on MCF-7 cell line as a model cancerous cell and HFF-2 cell line a model body cell.
- In-vivo cytotoxicity analyses on BALB/c mice at diffirent nanoparticle concentration.

## Time Table

| No. | Activity                             | Time needed (month) |  |  |
|-----|--------------------------------------|---------------------|--|--|
| 1   | Literature review                    | 2                   |  |  |
| 2   | Buy supplies and materials           | 2                   |  |  |
| 3   | Synthesis and modification           | 3                   |  |  |
| 4   | Characterization                     | 5                   |  |  |
| 5   | Performance test                     | 5                   |  |  |
| 6   | Final experiments and thesis writing | 2                   |  |  |

Thanks for your attention

Table 1.1 Composition of RHA derived from calcination of raw RH at 600 °C for 12 h.<sup>29</sup>

| SiO <sub>2</sub> | $Al_2O_3$ | Fe <sub>2</sub> O <sub>3</sub> | CaO  | MgO  | K <sub>2</sub> O | Na <sub>2</sub> O | MnO  | TiO <sub>2</sub> | $P_2O_5$ |
|------------------|-----------|--------------------------------|------|------|------------------|-------------------|------|------------------|----------|
| 93.2             | 0.13      | 0.07                           | 1.23 | 0.25 | 0.78             | 0.08              | 0.33 | 0.006            | 0.15     |